

## Novel Naphthoquinones from *Conospermum incurvum*

Jin-Rui Dai, Laurent A. Decosterd, Kirk R. Gustafson,  
John H. Cardellina II, Glenn N. Gray, and Michael R. Boyd

*J. Nat. Prod.*, **1994**, 57 (11), 1511-1516 • DOI:  
10.1021/np50113a006 • Publication Date (Web): 01 July 2004

Downloaded from <http://pubs.acs.org> on April 4, 2009

### More About This Article

---

The permalink <http://dx.doi.org/10.1021/np50113a006> provides access to:

- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article



**ACS Publications**  
High quality. High impact.

Journal of Natural Products is published by the American  
Chemical Society, 1155 Sixteenth Street N.W., Washington,  
DC 20036

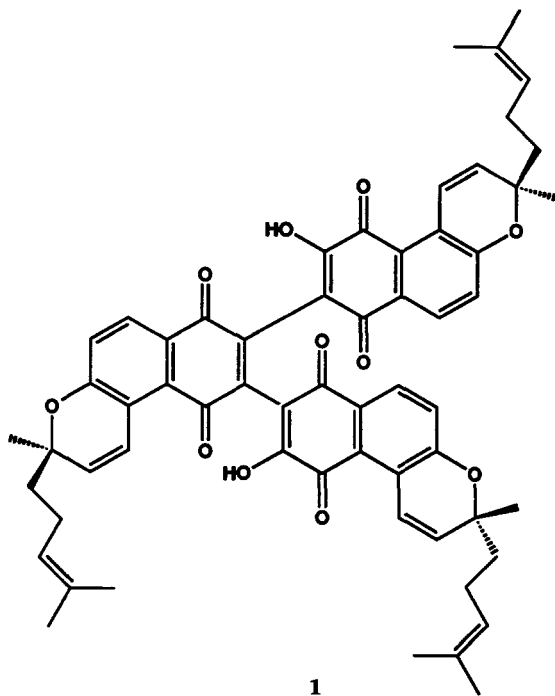
## NOVEL NAPHTHOQUINONES FROM *CONOSPERMUM INCURVUM*

JIN-RUI DAI,<sup>1</sup> LAURENT A. DECOSTERD,<sup>2</sup> KIRK R. GUSTAFSON, JOHN H. CARDELLINA II,  
GLENN N. GRAY, and MICHAEL R. BOYD\*

Laboratory of Drug Discovery Research and Development, Developmental Therapeutics Program,  
Division of Cancer Treatment, National Cancer Institute, Building 1052, Room 121,  
Frederick, Maryland 21702-1201

**ABSTRACT.**—During the reisolation of the trimeric naphthoquinone derivative conocurvone [1] from an extract of the Australian shrub *Conospermum incurvum*, six monomeric naphthoquinones were isolated. These include three novel 1,4-naphthoquinone derivatives: 3-methyl-14,15-dihydro-15-hydroxyteretifolione B [3], 3-methyl-14,15-dihydro-15-hydroxyteretifolione B methyl ether [4], and 2,3-dimethyl-6-hydroxy-7-methoxy-1,4-naphthoquinone [5]. In addition, the previously reported compounds 3-methylteretifolione B [6], 3-methylteretifolione B methyl ether [7], and 8-geranyl-2,7-dihydroxy-3-methyl-1,4-naphthoquinone [8] were isolated and identified. The structures of the novel 1,4-naphthoquinones were elucidated by spectral methods. While conocurvone [1] is a potent inhibitor of HIV-1-induced cell killing, all of the monomeric naphthoquinone derivatives were inactive against HIV-1.

A previous report from our laboratory detailed the isolation of conocurvone [1], a novel HIV-1 inhibitory trimeric naphthoquinone from an extract of the endemic Australian shrub *Conospermum incurvum* Lindley (Proteaceae) (1). The related monomeric quinone teretifolione B [2] was also obtained and its absolute stereochemistry estab-



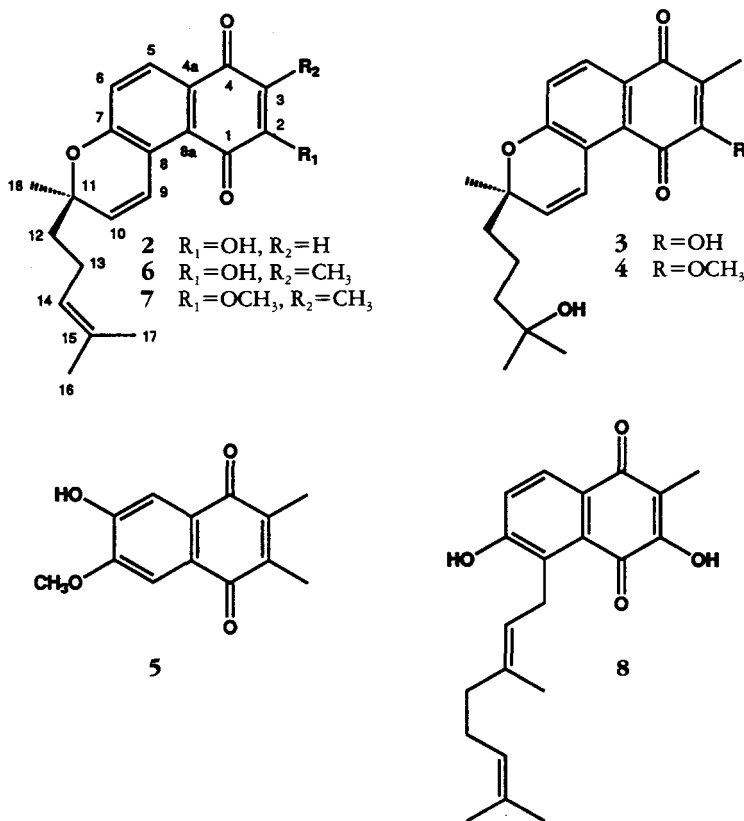
<sup>1</sup>Chemical Synthesis and Analysis Laboratory, Program Resources, Inc./DYNCORP, Frederick, Maryland 21702-1201.

<sup>2</sup>Current Address: Division of Clinical Pharmacology, Department of Internal Medicine, Centre Universitaire Hospitalier Vaudois, Lausanne, Switzerland.

lished via an X-ray crystallographic analysis. We have been investigating the reisolation of conocurvone [1] on a larger scale (2) to support detailed biological evaluation of its anti-HIV-1 activity. The crude extract was initially fractionated on a Sanki centrifugal partition chromatograph (cpc) in the ascending mode with hexane-EtOAc-MeOH-H<sub>2</sub>O (17:7:13:3). Fractions from the cpc were further purified by hplc or by reinjection on the Sanki cpc to afford conocurvone [1], teretifolione B [2], and six additional naphthoquinones. Three of these were novel 1,4-naphthoquinones, namely, 3-methyl-14,15-dihydro-15-hydroxyteretifolione B [3], the related methyl ether [4], and 2,3-dimethyl-6-hydroxy-7-methoxy-1,4-naphthoquinone [5].<sup>3</sup> The other three compounds, 3-methylteretifolione B [6], 3-methylteretifolione B methyl ether [7], and 8-geranyl-2,7-dihydroxy-3-methyl-1,4-naphthoquinone [8]<sup>3</sup> have previously been reported from *C. teretifolium* (3,4). The structures of compounds 3–5 were elucidated by spectroscopic techniques, while the structures of 6–8 were confirmed by independent spectral analyses and comparison of their spectral data with published values.

### RESULTS AND DISCUSSION

The presence of conocurvone [1] and teretifolione B [2] was confirmed by comparison with authentic samples. The spectral data obtained in this investigation for 3-methylteretifolione B [6], 3-methylteretifolione B methyl ether [7], and 8-geranyl-2,7-dihydroxy-3-methyl-1,4-naphthoquinone [8] were consistent with literature values



<sup>3</sup>For clarity of discussion and comparison of spectral data, the numbering scheme for this compound has been assigned so that it is consistent with that of teretifolione B [2] and others in the series of monomeric naphthoquinones.

(3,4). However, only  $^1\text{H}$ -nmr data have previously been reported for these compounds. We have independently confirmed the proposed structures and fully assigned the  $^1\text{H}$ - and  $^{13}\text{C}$ -nmr resonances of these compounds by a variety of nmr techniques, including HMQC and HMBC proton-detected heteronuclear correlation experiments. The  $^1\text{H}$ -nmr data obtained for compounds **6** and **7** were in very close agreement with published values, but a slight discrepancy was noted for compound **8**. The C-10 olefin proton was observed at  $\delta$  5.13, while the previously reported value was  $\delta$  5.25 (3,4). All other proton resonances that were measured for **8** were similar to literature values. An nOe experiment showing significant nOe interactions between the C-10 proton and the C-12 allylic protons confirmed the olefin geometry as *E*. Therefore, we are unable to explain the minor difference between our data for **8** and the literature values (3,4).

Compound **3** was isolated as an optically active,  $[\alpha]_{\text{D}} + 38.8^\circ$ , dark orange powder. Hreims revealed a molecular formula of  $\text{C}_{21}\text{H}_{24}\text{O}_5$ , with a molecular ion at  $m/z$  356.1639. Absorptions in the ir ( $1644\text{ cm}^{-1}$ ) and uv (219,292 nm) spectra of **3** indicated the presence of conjugated carbonyl groups. The  $^1\text{H}$ - and  $^{13}\text{C}$ -nmr spectra (Tables 1 and 2) were closely related to those observed for 3-methylteretifolione B [**6**]. However, in the carbon spectrum of **3**, signals for the trisubstituted olefin previously assigned in **6** were replaced by an aliphatic methylene and a tertiary alcohol. Two exchangeable proton resonances at  $\delta$  7.44 and  $\delta$  4.07<sup>4</sup> indicated the presence of one phenol and one aliphatic

TABLE 1.  $^1\text{H}$ -Nmr Spectral Data (in  $\text{CDCl}_3$ ) for Compounds **3-8**.<sup>a</sup>

Position	Compound					
	<b>3</b>	<b>4</b>	<b>5</b>	<b>6</b>	<b>7</b>	<b>8</b>
H-5	7.95 d (8.5)	7.89 d (8.5)	7.53 s	7.93 d (8.5)	7.86 d (8.5)	7.99 d (8.5)
H-6	7.03 dd (8.5, 1.0)	6.99 dd (8.5, 1.0)		7.02 d (8.5)	6.96 dd (8.5, 1.0)	7.11 d (8.5)
H-8			7.49 s			
H-9	7.81 d (10.5)	7.71 d (10.5)		7.79 d (10.5)	7.69 d (10.5)	4.05 d (7.0, 2H)
H-10	5.92 d (10.5)	5.86 d (10.5)		5.90 d (10.5)	5.83 d (10.5)	5.13 m
H-12	1.70 m, 1.75 m	1.64 m, 1.74 m		1.66 m, 1.75 m	1.64 m, 1.72 m	2.05 m (2H)
H-13	1.47 m (2H)	1.50 m (2H)		2.07 m (2H)	2.04 m (2H)	2.05 m (2H)
H-14	1.44 m (2H)	1.43 m (2H)		5.04 m	5.02 m	5.01 m
CH <sub>3</sub> -16	1.17 s	1.16 s		1.61 s	1.60 s	1.63 s
CH <sub>3</sub> -17	1.18 s	1.17 s		1.52 s	1.51 s	1.56 s
CH <sub>3</sub> -18	1.41 s	1.38 s		1.41 s	1.39 s	1.84 s
CH <sub>3</sub> -2			2.11 s			
CH <sub>3</sub> -3	2.04 s	2.02 s	2.11 s	2.02 s	1.99 s	2.04 s
OCH <sub>3</sub> -2		4.00 s			3.97 s	
OCH <sub>3</sub> -7			4.01 s			
OH	4.07 <sup>b</sup> s 7.44 s	4.07 <sup>b</sup> s	6.10 s	7.49 s		7.49 s

<sup>a</sup>Chemical shifts ( $\delta$ ) were referenced relative to the internal solvent signal. Coupling constants in parentheses are in Hz.

<sup>b</sup>Observed in  $\text{DMSO}-d_6$ .

<sup>4</sup>Exchangeable proton resonances were observed in  $\text{DMSO}-d_6$  spectra.

TABLE 2.  $^{13}\text{C}$ -Nmr Spectral Data (in  $\text{CDCl}_3$ ) for Compounds 3–8.<sup>a</sup>

Position	Compound					
	3	4	5	6	7	8
C-1	183.2	183.6	184.2	183.2	183.1	183.0
C-2	153.2	158.4	142.8 <sup>b</sup>	153.3	158.3	153.5
C-3	118.6	130.4	142.9 <sup>b</sup>	118.6	120.5	118.4
C-4	184.4	185.1	184.3	184.4	185.0	184.7
C-4a	126.8	126.3	127.7	126.9	126.2	127.8
C-5	128.7	128.3	112.0	128.7	128.2	127.6
C-6	121.7	121.0	150.1	121.7	120.8	121.0
C-7	157.8	158.5	150.4	158.0	158.5	159.4
C-8	121.2	120.6	107.6	121.2	130.3	130.3
C-8a	123.2	126.0	126.6	123.3	126.0	127.7
C-9	120.2	120.4		120.1	120.2	25.3
C-10	135.1	134.0		135.1	140.0	120.2
C-11	79.2	79.2		79.2	79.1	139.2
C-12	41.6	41.6		41.2	41.1	39.7
C-13	18.7	18.8		22.6	22.6	26.4
C-14	43.6	43.8		123.5	123.5	123.7
C-15	70.9	70.9		132.1	132.0	132.0
C-16	29.3 <sup>c</sup>	29.3 <sup>c</sup>		25.6	25.6	25.6
C-17	29.2 <sup>c</sup>	29.2 <sup>c</sup>		17.6	17.5	17.7
C-18	26.5	26.5		26.5	26.5	16.4
CH <sub>3</sub> -2			12.8			
CH <sub>3</sub> -3	8.5	9.1	12.8	8.5	9.0	8.5
OCH <sub>3</sub>		60.8	56.5		60.7	

<sup>a</sup>Chemical shifts ( $\delta$ ) were referenced relative to the internal solvent signal.

<sup>b,c</sup>Assignments within a column may be reversed.

hydroxyl group. This suggested that compound **3** was the hydrate of **6** resulting from Markovnikov-type addition of  $\text{H}_2\text{O}$  to the C-14–C-15 side-chain olefin. Long-range heteronuclear correlations from the oxygenated C-15 carbon ( $\delta$  70.9) to the C-13 methylene protons and from C-14 ( $\delta$  43.6) to the C-16, C-17 methyl groups supported this assignment. The C-2 hydroxyl and C-3 methyl substitution pattern about the naphthoquinone moiety was confirmed by correlations from C-1 ( $\delta$  183.2) to the C-2 hydroxyl proton and from C-4 ( $\delta$  184.4) to H-5. All other heteronuclear correlations observed were fully consistent with the proposed structure of 3-methyl-14,15-dihydro-15-hydroxyteretifolione B for **3**, and permitted complete assignments of all  $^1\text{H}$ - and  $^{13}\text{C}$ -nmr signals to be made. The fact that both teretifolione B [**2**] and compound **3** have strong, positive optical rotations indicates that they have the same (*R*) absolute stereochemistry at C-11.

Compound **4**,  $[\alpha]_{\text{D}} + 32.1^\circ$ , was also obtained as a dark orange powder. The hreims of **4** revealed a molecular formula of  $\text{C}_{22}\text{H}_{26}\text{O}_5$  ( $M^+$ ,  $m/z$  370.1805) and the ir and uv spectra indicated the presence of conjugated carbonyl chromophores (1698, 1666  $\text{cm}^{-1}$ ; 222, 266 nm). The  $^1\text{H}$ - and  $^{13}\text{C}$ -nmr spectra of **4** (Tables 1 and 2) closely matched those of compound **3**. However, there was only one exchangeable proton signal ( $\delta$  4.07),<sup>4</sup> but there was an additional methoxyl group present ( $^1\text{H}$   $\delta$  4.00 3H, s;  $^{13}\text{C}$   $\delta$  60.8). HMBC correlations from C-2 ( $\delta$  158.4) to the methoxyl protons and from C-4 ( $\delta$  185.1) to the C-3 methyl protons defined the regiochemical placement of these substituents. Complete  $^1\text{H}$ - and  $^{13}\text{C}$ -nmr assignments for compound **4** were made from heteronuclear correlation data. These data established that compound **4** is the C-2 methyl ether derivative of compound **3**.

Compound **5** was obtained as an optically inactive yellow powder. The hreims spectrum of **5** indicated a molecular formula of  $C_{13}H_{12}O_4$  for the molecular ion at  $m/z$  232.0729. The ir and uv spectra of **5** indicated the presence of conjugated carbonyl groups (ir  $1644\text{ cm}^{-1}$ ; uv 277 nm). The  $^1\text{H}$ -nmr spectrum of **5** (Table 1) exhibited two singlet aromatic protons at  $\delta$  7.53 and 7.49, a six-proton singlet due to two methyl groups at  $\delta$  2.11, one methoxyl at  $\delta$  4.01 (s) and an exchangeable phenol resonance at  $\delta$  6.10. The  $^{13}\text{C}$ -nmr spectrum (Table 2) displayed a two-carbon signal at  $\delta$  12.8 due to two methyls, one methoxyl group at  $\delta$  56.5 ppm, two quinone carbonyls at  $\delta$  184.2 and 184.3, and eight additional  $\text{sp}^2$  carbons making up a naphthoquinone skeleton. This indicated that compound **5** was a 1,4-naphthoquinone substituted with two methyls, one methoxyl and one hydroxyl group.

The assignment of  $^1\text{H}$ - and  $^{13}\text{C}$ -nmr spectral data and the structural elucidation of compound **5** were supported by HMQC and HMBC experiments. Long-range coupling observed between the quinone carbonyl carbons and the methyl protons established the methyl substituents at C-2 and C-3. Correlations between H-5 ( $\delta$  7.53), C-4 ( $\delta$  184.3), C-7 ( $\delta$  150.4), and C-8a ( $\delta$  126.6), between H-8 ( $\delta$  7.49), C-6 ( $\delta$  150.1), and C-4a ( $\delta$  127.7), between the methoxy group ( $\delta$  4.01) and C-7 ( $\delta$  150.4), as well as between the phenolic OH ( $\delta$  6.10) and C-5 ( $\delta$  112.0) and C-7 ( $\delta$  150.4), firmly established the structure of compound **5** as 2,3-dimethyl-6-hydroxy-7-methoxy-1,4-naphthoquinone.

Testing of **3–8** in an in vitro XTT-based anti-HIV-1 assay (5,6), revealed that none of these "monomeric" naphthoquinone derivatives were active. This is in sharp contrast to the reported potent anti-HIV activity of the "trimeric" conocurvone [**1**], but consistent with the reported inactivity of the monomer teretifolione B [**2**] and a related 1,4-naphthoquinone "dimer" (1).

## EXPERIMENTAL

**GENERAL EXPERIMENTAL PROCEDURES.**—Centrifugal partition chromatography (cpc) was performed at  $24^\circ$  with a cartridge Sanki model NMF instrument. An ISCO V absorbance detector (280 nm) and a Foxy fraction collector were connected to the main Sanki cpc unit. Hplc was performed on a Rainin system equipped with Rabbit-HP HPX pumps and a Dynamax UV-1 detector. Optical rotations were measured on a Perkin-Elmer 241 polarimeter; uv spectra were recorded on a Beckman DU-64 spectrophotometer; ir spectra were recorded on a Perkin-Elmer 1600 series Ftir; nmr spectra were recorded on a Varian VXR 500 spectrometer at  $25^\circ$ . Hreims spectra were recorded with a Finnigan MAT 90 spectrometer.

**PLANT MATERIAL.**—The roots of *Conospermum incurvum* Lindley (collection number Q73H128) were collected in February 1991, on the Northern Sand Plains, Western Australia. Ground plant material (11.416 kg) was percolated at room temperature in  $\text{CH}_2\text{Cl}_2$ , and then in  $\text{CH}_2\text{Cl}_2$ -MeOH (1:1), and finally washed with 100% MeOH. Solvent was removed *in vacuo* to produce  $\text{CH}_2\text{Cl}_2$ ,  $\text{CH}_2\text{Cl}_2$ -MeOH (1:1), and MeOH extracts (69.81 g, 145.73 g, and 23.81 g, respectively).

**EXTRACTION AND ISOLATION.**—A 1.5-g portion of the  $\text{CH}_2\text{Cl}_2$  extract was fractionated by cpc with hexane-EtOAc-MeOH- $\text{H}_2\text{O}$  (17:7:13:3), using the ascending mode, a spin rate of 500 rpm, a flow rate of 2.7 ml/min, and uv detection at 280 nm. A total of 12 fractions were obtained. Conocurvone [**1**] (9 mg) and compound **6** (30 mg) were obtained from cpc fraction 3 by two different hplc methods. Pure conocurvone [**1**] was obtained by phenyl-bonded phase hplc using  $\text{CH}_3\text{CN}$ - $\text{H}_2\text{O}$  (17:3, 0.1% HOAc by volume), while compound **6** was purified by Si gel hplc eluting with  $\text{CH}_2\text{Cl}_2$ -MeOH (19:1). Teretifolione B [**2**] (30 mg) was isolated from cpc fraction 5, compound **4** (20 mg) was isolated from cpc fraction 7 (68 mg), and compound **8** (5 mg) was isolated from cpc fraction 6 (24 mg) using the same hplc conditions as those used for **6**. Hplc on a phenyl-bonded phase, eluting with  $\text{CH}_3\text{CN}$ - $\text{H}_2\text{O}$  (3:1, 0.1% HOAc by volume), of 23 mg of cpc fraction 9 and 40 mg of cpc fraction 8, provided compound **3** (12 mg) and compound **5** (7 mg), respectively. A 100-mg portion of cpc fraction 2 was reinjected on the Sanki cpc using the exact same conditions as the original cpc run to give 60 mg of compound **7**.

**3-Methyl-14,15-dihydro-15-hydroxyteretifolione B** [**3**].— $[\alpha]_D +38.8^\circ$  ( $c=0.56$ ,  $\text{CHCl}_3$ ); uv  $\lambda$  max (MeOH) (log  $\epsilon$ ) 219 (5.33), 292 (5.12), 392 (4.66) nm; ir (film)  $\nu$  max 3373, 2970, 1644, 1561, 1391, 1351, 847  $\text{cm}^{-1}$ ; for  $^1\text{H}$  and  $^{13}\text{C}$  nmr, see Tables 1 and 2, respectively; eims  $m/z$  356 ( $\text{M}^+$ , 4), 341  $[\text{M}-\text{CH}_3]^+$  (2), 338  $[\text{M}-\text{H}_2\text{O}]^+$  (1), 270  $[\text{M}-\text{C}_5\text{H}_{10}\text{O}]^+$  (1), 255  $[\text{M}-\text{C}_5\text{H}_{10}\text{O}-\text{CH}_3]^+$  (100), 227

$[M-C_5H_{10}O-CH_3-CO]^+$  (19), 199 (1), 149 (4), 85 (1), 69 (1); hreims  $m/z$  356.1639 ( $M^+$ ,  $C_{21}H_{24}O_5$ , calcd 356.1624).

*3-Methyl-14,15-dihydro-15-hydroxyteretifolione B methyl ether* [4].— $[\alpha]_D + 32.1^\circ$  ( $c=0.80$ ,  $CHCl_3$ ); uv  $\lambda$  max (MeOH) (log  $\epsilon$ ) 222 (5.23), 266 (5.12), 370 (4.41) nm; ir (film)  $\nu$  max 3478, 2970, 1698, 1666, 1564, 1372, 1279, 983  $cm^{-1}$ ; for  $^1H$  and  $^{13}C$  nmr, see Tables 1 and 2, respectively; eims  $m/z$  370 ( $M^+$ , 6), 355  $[M-CH_3]^+$  (5), 352  $[M-H_2O]^+$  (4), 297 (2), 285 (11), 284  $[M-C_5H_{10}O]^+$  (1), 270 (33), 269  $[M-C_5H_{10}O-CH_3]^+$  (100), 241  $[M-C_5H_{10}O-CH_3-CO]^+$  (1), 226 (12), 170 (4), 149 (5), 115 (2), 83 (1), 59 (2); hreims  $m/z$  370.1805 ( $M^+$ ,  $C_{22}H_{26}O_5$ , calcd 370.1780).

*2,3-Dimethyl-6-hydroxy-7-methoxy-1,4-naphthoquinone* [5].—Uv  $\lambda$  max (MeOH) (log  $\epsilon$ ) 277 (5.45), 352 (4.28) nm; ir (film)  $\nu$  max 3300–3400, 1644, 1580, 1524, 1356, 1313, 1200, 889  $cm^{-1}$ ; for  $^1H$  and  $^{13}C$  nmr, see Tables 1 and 2, respectively; eims  $m/z$  232 ( $M^+$ , 100), 217  $[M-CH_3]^+$  (6), 215 (4), 214  $[M-H_2O]^+$  (1), 204  $[M-CO]^+$  (42), 189  $[M-CO-CH_3]^+$  (57), 161 (17), 150 (11), 133 (5), 122 (13), 115 (5), 83 (3), 63 (2), 51 (6); hreims  $m/z$  232.0729 ( $M^+$ ,  $C_{13}H_{12}O_4$ , calcd 232.0736).

*3-Methylteretifolione B* [6].— $[\alpha]_D + 29.7^\circ$  ( $c=0.33$ ,  $CHCl_3$ ); for  $^1H$  and  $^{13}C$  nmr, see Tables 1 and 2, respectively; eims  $m/z$  338 ( $M^+$ , 14), 323  $[M-CH_3]^+$  (4), 295  $[M-CH_3-CO]^+$  (1), 270 (1), 269  $[M-C_6H_9]^+$  (2), 255  $[M-C_6H_{11}]^+$  (100), 227  $[M-C_6H_{11}-CO]^+$  (29), 199 (1), 115 (3), 69 (5).

*3-Methylteretifolione B methyl ether* [7].— $[\alpha]_D + 37.3^\circ$  ( $c=0.73$ ,  $CHCl_3$ ); for  $^1H$  and  $^{13}C$  nmr, see Tables 1 and 2, respectively; eims  $m/z$  352 ( $M^+$ , 49), 337  $[M-CH_3]^+$  (16), 309  $[M-CH_3-CO]^+$  (4), 284 (1), 283  $[M-C_6H_9]^+$  (3), 269  $[M-C_6H_{11}]^+$  (100), 255 (7), 241  $[M-C_6H_{11}-CO]^+$  (4), 226 (24), 213 (2), 182 (13), 144 (6), 115 (6), 77 (2), 69 (15).

*8-Geranyl-2,7-dihydroxy-3-methyl-1,4-naphthoquinone* [8].—For  $^1H$  and  $^{13}C$  nmr, see Tables 1 and 2, respectively; eims  $m/z$  340 ( $M^+$ , 71), 322  $[M-H_2O]^+$  (4), 297  $[M-C_3H_7]^+$  (54), 285  $[M-C_4H_9]^+$  (2), 271  $[M-C_6H_9]^+$  (62), 257  $[M-C_6H_{11}]^+$  (22), 253 (68), 229  $[M-C_6H_{11}]^+$  (100), 218 (66), 187 (44), 149 (37), 123 (35), 115 (18).

ANTI-HIV EVALUATIONS.—DMSO solutions of the purified compounds were tested in the XTT-based in vitro anti-HIV assay as described elsewhere (5,6).

#### ACKNOWLEDGMENTS

We thank T. McCloud for extraction of the plant material and J.B. McMahon for the anti-HIV evaluations. L.A. Decosterd was supported by a postdoctoral fellowship from the Swiss Cancer League.

#### LITERATURE CITED

1. L.A. Decosterd, I.C. Parsons, K.R. Gustafson, J.H. Cardellina, II, J.B. McMahon, G.M. Cragg, Y. Murata, L.K. Pannell, J.R. Steiner, J. Clardy, and M.R. Boyd, *J. Am. Chem. Soc.*, **115**, 6673 (1993).
2. J.-R. Dai, L.A. Decosterd, K.R. Gustafson, J.H. Cardellina, II, and M.R. Boyd, *J. Liq. Chromatogr.*, submitted (1994).
3. J.R. Cannon, K.R. Joshi, I.A. McDonald, R.W. Retallack, A.F. Sierakowski, and L.C.H. Wong, *Tetrahedron Lett.*, **32**, 2795 (1975).
4. R.H. Thomson, "Naturally Occurring Quinones III: Recent Advances," Chapman and Hall, New York, 1987, pp. 206–209.
5. M.R. Boyd, in: "AIDS Etiology, Diagnosis, Treatment and Prevention." Ed. by V.T. DeVita, S. Hellman, and S.A. Rosenberg, Lippincot, Philadelphia, 1988, pp. 305–319.
6. R.J. Gulakowski, J.B. McMahon, P.G. Staley, R.A. Moran, and M.R. Boyd, *J. Virol. Methods*, **33**, 87 (1991).

Received 10 May 1994